

## INFORMATION DISCLOSURE CITATION WITH DOCUMENT COPIES

Submitted by:  General Motors Corporation	Atty. Docket No. GP-302579	Application No. 10/824,876
	Applicant: Meisner et al.	Confirmation No. 7304
	Filing Date: April 15, 2004	Art Unit: 1754

U.S. PATENT DOCUMENTS				
Examiners Initials	Document Number	Date	Name	Classification Class/Subclass
	4,007,257	02-1977	Lemieux et al.	423/646
	6,015,041	01-2000	Heung	206/70
	6,159,538	12-2000	Rodriguez et al.	427/213.31
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	6,329,076	12-2001	Kawabe et al.	428/656
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	2003/0113252	06-2003	Chen et al.	423/414
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FOREIGN PATENT DOCUMENTS					
Examiners Initials	Document Number	Date	Country	Classification Class/Subclass	Translation Yes   No

Examiner:

Date Considered:

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NON-PATENT DOCUMENTS		
Examiners Initials		Include as applicable: Author, Title, Date, Publisher, Edition/Volume, Pertinent Pages
	1	Cenzual et al., "Inorganic Structure Types with Revised Space Group", Acta Cryst., Vol. B47 (1991) 433-439.
	2	Chen et al., "Hydrogen Storage in Metal Nitride Systems", Edited by Ricardo B. Schwartz, Symposium V, Materials for Energy Storage, Generation and Transport, Vol. 730 (April 2-4, 2002) 376 and 385.
	3	Chen et al., "Interaction of Hydrogen with Metal Nitrides and Imides", Nature Publishing Group [Vol. 420] (November 21, 2002) 302-304 with Supplement pp. 1-6
	4	Goubeau, et al., "Über ternäre Metall-Bornitride", Zeitschrift für anorganische und allgemeine Chemie, Vol. 310 (1961) 248-260.
	5	Hu et al., "Ultrafast Reaction between LiH and NH <sub>3</sub> during H <sub>2</sub> Storage in Li <sub>3</sub> N"; J. Phys. Chem. A; Vol. 107, No. 46 (November 20, 2003) 9737-9739.
	6	Ichikawa et al., "Mechanism of Novel Reaction for LiNH and LiH to Li <sub>2</sub> NH and H <sub>2</sub> as a Promising Hydrogen Storage System"; J. Phys. Chem. B; Vol. 108, No. 23 (May 5, 2004) 7887-7892.
	7	Jacobs et al., "Preparations and Properties of Magnesium Amide and Imide", Journal for Anorganic and General Chemistry, Band [Vol.] 870 (1969) 254-261. (English translation only; original German not available).
	8	JCPDS X-Ray Database; pattern no. 00-007-0245 – Li <sub>3</sub> AlN <sub>2</sub>
	9	JCPDS X-Ray Database; pattern no. 00-036-1016 – β-Mg <sub>3</sub> B <sub>2</sub> N <sub>4</sub>
	10	JCPDS X-Ray Database; pattern no. 00-042-0868 – Mg <sub>3</sub> BN <sub>3</sub>
	11	JCPDS X-Ray Database; pattern no. 00-044-1497 – Mg <sub>3</sub> BN <sub>3</sub>
	12	JCPDS X-Ray Database; pattern no. 16-273 – Li <sub>3</sub> BN <sub>2</sub>
	13	JCPDS X-Ray Database; pattern no. 40-1166 – Li <sub>3</sub> BN <sub>2</sub>
	14	JCPDS X-Ray Database; pattern no. 80-2274 – Li <sub>3</sub> BN <sub>2</sub>
	15	Juza et al., "Die ternären Nitride Li <sub>3</sub> AlN <sub>2</sub> und Li <sub>3</sub> GaN <sub>2</sub> "; Zeitschrift für Anorganische Chemie, Vol. 257 (1948) 13-25.
	16	Juza et al., "Metal amides and metal nitrides", 25 <sup>th</sup> Part, Journal for Anorganic and General Chemistry, 1951 Volume 266, 325-330. (English translation and German language document).
	17	Pinkerton et al., "Hydrogen Desorption Exceeding Ten Weight Percent from the New Quaternary Hydride Li <sub>3</sub> BN <sub>2</sub> H <sub>8</sub> " ACS Publications, <a href="http://pubs.acs.org/cgi-bin/abstract.cgi/jpcbfk/2005/109/i01/abs/jp0455475.html">http://pubs.acs.org/cgi-bin/abstract.cgi/jpcbfk/2005/109/i01/abs/jp0455475.html</a>
	18	Pinkerton et al., "Bottling the Hydrogen Genie", The Industrial Physicist, (February/March 2004) 20-23.

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	19	Villars et al., "ASM International Handbook of Ternary Alloy Phase Diagrams", Al Li N; $AlLi_3N_2$ (1) Crystallographic Data (1997).
	20	Villars et al., "ASM International Handbook of Ternary Alloy Phase Diagrams", B Li N; $BLi_3N_2$ (LT) (2) Crystallographic Data (1997).
	21	Villars et al., "ASM International Handbook of Ternary Alloy Phase Diagrams", B Li N; $BLi_3N_2$ (HT) (2) Crystallographic Data (1997).
	22	Villars, P., "Pearson's Handbook Desk Edition", Crystallographic Data for Intermetallic Phases, Ac - $Cr_2Se_4Zr$ , Vol. 1, p. 416 (1997) 771 and 776.
	23	Yamane et al., "High- and Low-Temperature Phases of Lithium Boron Nitride, $Li_3BN_2$ Preparation, Phase Relation, Crystal Structure, and Ionic Conductivity", J. Solid State Chemistry, Vol. 71, 1987) 1-11.
	24	Yamane et al., "Structure of a New Polymorph of Lithium Boron Nitride, $Li_3BN_2$ ", J. Solid State Chemistry, Vol. 65, (1986) 6-12.

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